

171. Synthesis of 1,2,5-Thiadiazepine Derivatives by Ring Enlargement of 1,2-Thiazetidin-3-one 1,1-Dioxides with 3-Amino-2*H*-azirines

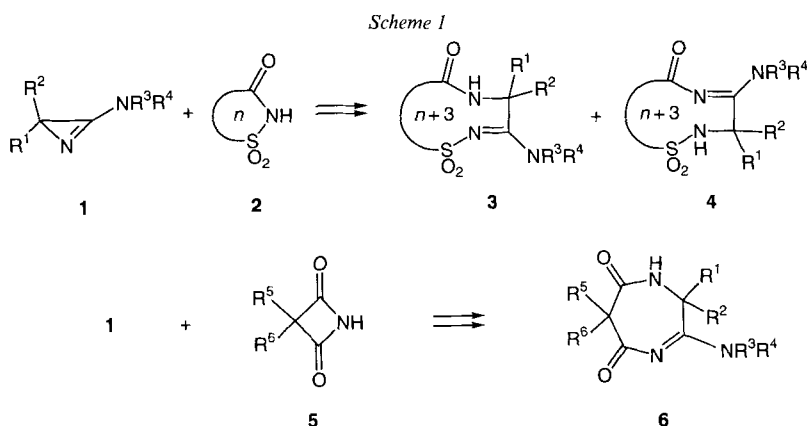
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(3.IX.96)

At 0° in MeCN, 2,2-disubstituted 3-amino-2*H*-azirines **1** and 4,4-disubstituted 1,2-thiazetidin-3-one 1,1-dioxides **7** react smoothly to give 1,2,5-thiadiazepin-6-one 1,1-dioxides of type **8** (Scheme 2). The reaction mechanism of this regioselective ring enlargement to seven-membered heterocycles follows previously described pathways. The structures of **7a** and **8b** were established by X-ray crystallography (see Figs. 1 and 2).

Introduction. – In several papers, we have shown that 3-amino-2*H*-azirines **1** and NH-acidic heterocycles with $pK_a < 8$ react *via* ring enlargement to give medium-sized heterocycles [1–7]. Whereas in some cases the primary products could not be isolated because of their further rearrangement under the reaction conditions [4–6], they were obtained as stable compounds in reactions with five- to seven-membered sulfonamides of type **2** (Scheme 1). A smooth reaction at 0–20° was observed with saccharin [8] [9] and other 1,2-thiazol-3-one 1,1-dioxides [2], yielding 1,2,5-thiadiazocine derivatives **3** ($n = 5$, Scheme 1). In one example, the isomeric heterocycle of type **4** ($n = 5$) was formed as a minor product [2]. With analogous six-membered derivatives, a 1,2,4-thiadiazin-3-one 1,1-dioxide [4] and a 1,2-thiazin-3-one 1,1-dioxide [3], the transformation was slower, and with a seven-membered 1,2-thiazepin-3-one 1,1-dioxide, the reaction was sluggish even at 50–65° [7]. In the latter case, yields of ten-membered heterocycles **3** ($n = 7$) were low, and the isomer of type **4** ($n = 7$) was always formed as a by-product.



¹) Part of the planned Ph.D. thesis of T. R. M.

The four-membered malonimides **5** (= azetidine-2,4-diones) also reacted with **1**, in propan-2-ol at room temperature, to give 1,4-diazepine-5,7-dione derivatives **6** [10] (Scheme 1). Therefore, we expected that the corresponding 1,2-thiazetid-3-one 1,1-dioxides **7** with **1** should undergo a ring enlargement to yield seven-membered heterocycles.

Results and Discussion. – Several years ago, some 1,2-thiazetid-3-one 1,1-dioxides **7** were synthesized from the corresponding 2,2-disubstituted 2-(chlorosulfonyl)acetyl chlorides in liquid NH₃ [11] [12]. Following the same protocol, we prepared the 4,4-dimethyl and the 4,4-diethyl derivative (**7a** and **7b**, respectively). These relatively acidic compounds²⁾ were obtained as colorless, crystalline materials. In spite of the high ring strain, which is reflected by the very acute and obtuse angles in the crystal structure of **7a** (Fig. 1), the compounds are remarkably stable (cf. [11]).

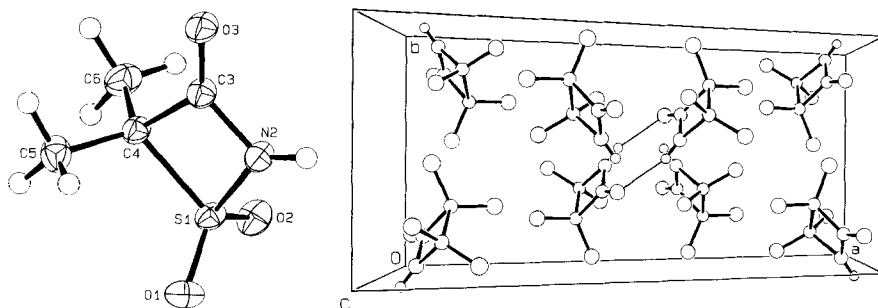


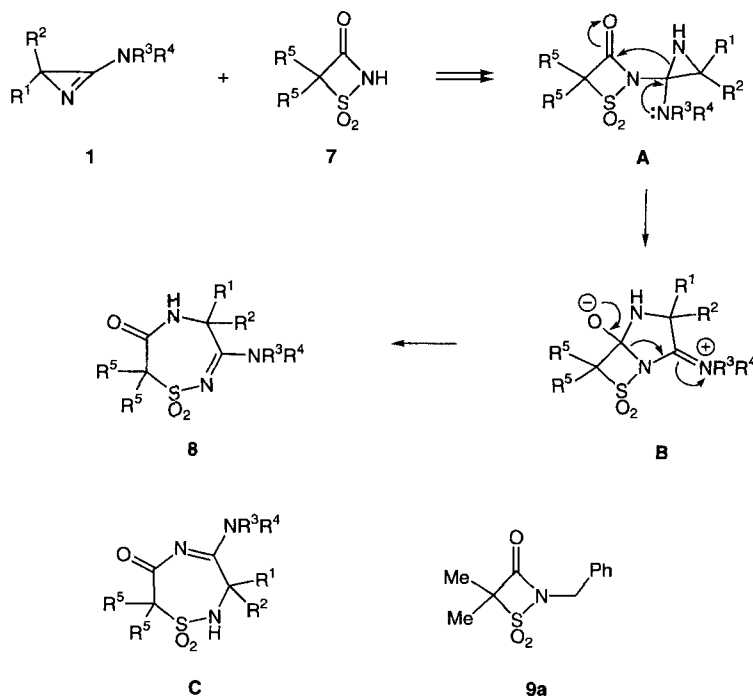
Fig. 1. ORTEP Plot [13] of the molecular structure of **7a** (displacement ellipsoids with 50% probability) and packing diagram

The four-membered ring of **7a** is almost planar with intraannular torsion angles between $-3.3(1)^\circ$ and $3.6(1)^\circ$. The carbonyl O-atom also deviates only slightly from the ring plane (S(1)–N(2)–C(3)–O(3) $-177.4(2)^\circ$, S(1)–C(4)–C(3)–O(3) $177.7(2)^\circ$), as does H–N(2). The Me groups at C(4) and the O-atoms of the SO₂ group are in a nearly perfect eclipsed orientation (O(1)–S(1)–C(4)–C(5) $0.1(1)^\circ$, O(2)–S(1)–C(4)–C(6) $8.4(1)^\circ$). The intraannular bond angle at the S-atom is $78.75(7)^\circ$, at the N-atom $96.3(1)^\circ$, at the sp³ C-atom $84.8(1)^\circ$, and at the carbonyl C-atom $100.0(1)^\circ$. The N(2)–C(3) bond is short ($1.366(2)$ Å); this indicates, together with the planarity of N(2), the delocalization of the lone electron pair of N(2) within the lactam group. The NH group of **7a** forms an intermolecular H-bond with the carbonyl group of a neighboring molecule (N···O distance $2.857(2)$ Å, N–H···O angle $158(2)^\circ$). The interactions link the molecules into dimeric units where the two molecules of the dimer are related by a center of inversion (Fig. 1).

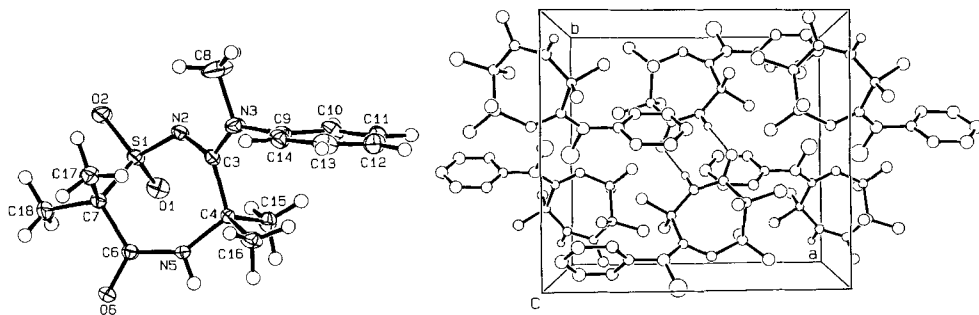
In MeCN at 0°, **7a** and **7b** reacted smoothly with 3-amino-2H-azirines **1a–d** to give a single 1:1 adduct of type **8** as a colorless, crystalline material. The 1,2,5-thiadiazepine structure (Scheme 2) was deduced from the spectral data, and in the case of **8b**, it was established by X-ray crystallography (Fig. 2). The yields of **8** are usually high (70–90% isolated material, Table 1). Only in the reactions with **1a**, the yields were lower, although the transformations proceeded rapidly.

²⁾ 1,2-Thiazetid-3-one 1,1-dioxides **7** are easily dissolved in aqueous NaHCO₃ solutions [11].

Scheme 2


 Table 1. Formation of 1,2,5-Thiadiazepin-6-one 1,1-Dioxides **8** from the Reaction of **1** and **7**

	R ¹	R ²	R ³	R ⁴		R ⁵	Product [%]
1a	Me	Me	Me	Me	7a	Me	8a (31)
1b	Me	Me	Me	Ph	7a		8b (91)
1c		-(CH ₂) ₄ -	Me	Ph	7a		8c (79)
1d	Me	i-Bu	Me	Ph	7a		8d (80)
1a	Me	Me	Me	Me	7b	Et	8e (60)
1b	Me	Me	Me	Ph	7b		8f (73)
1d	Me	i-Bu	Me	Ph	7b		8g (81)


 Fig. 2. ORTEP Plot [13] of the molecular structure of **8b** (displacement ellipsoids with 50% probability) and packing diagram

The seven-membered ring of **8b** is puckered with an almost planar (*Z*)-configured amide group (C(4)–N(5)–C(6)–C(7) 2.5(4)°, C(4)–N(5)–C(6)–O(6) –177.7(2)°). The amidine group is also nearly planar (N(2)–C(3)–N(3) 113.5(2)°, N(3)–C(3)–C(4) 120.6(2)°, N(2)–C(3)–C(4) 125.8(2)°), with the MeN group also being in this plane (N(2)–C(3)–N(3)–C(8) 4.4(3)°, C(4)–C(3)–N(3)–C(8) –179.0(2)°). The C–N bond lengths of the amidine group (N(2)–C(3) 1.308(2) Å, N(3)–C(3) 1.350(2) Å) also reflect a certain delocalization of the lone pair of N(3); but there is no conjugation with the Ph ring at N(3). The molecule forms an intermolecular H-bond between the NH group and the carbonyl O-atom of a neighboring molecule (N···O distance 2.894(2) Å, N–H···O angle 173(2)°). The H-bonds link the molecules into dimeric units in which the molecules of the dimer are related by a center of inversion (*Fig. 2*).

In analogy to previously reported reactions of **1** with NH-acidic compounds, we propose the mechanism depicted in *Scheme 2* for the formation of the seven-membered heterocycles **8**. After protonation of **1** by **7**, nucleophilic attack onto the amidinium C-atom yields aziridine **A**, which undergoes a ring enlargement to give the zwitterionic intermediate **B**. By a second ring enlargement, the latter rearranges to the final product **8**. An alternative reaction mechanism *via* direct nucleophilic addition of **1** to the carbonyl group of **7** can be excluded as no reaction took place under analogous conditions between **1a** and the *N*-benzylated derivative **9a**³⁾).

It is worth mentioning that no product of the isomeric structure **C** (*Scheme 2*) could be detected. Therefore, the ring enlargement of aziridine **A** occurred in a regioselective manner *via* nucleophilic attack of the aziridine N-atom at the carbonyl group leading to **B**. Ring enlargement *via* addition to the SO₂ group would yield products of type **C** (*cf.* discussion in [2] [7]). In intermolecular nucleophilic ring-opening reactions of derivatives of type **7** or **9**, cleavages of the N–C and/or N–S bond, *i.e.*, addition to the CO and SO₂ group, respectively, are observed. Hydrolysis of **7a** with diluted HCl yielded 1-carbamoyl-1-methylethylsulfonic acid (*via* SO₂ attack) and 2-sulfamoyl-2-methylpropanamide (*via* CO attack) in nearly equal amounts, whereas under similar conditions 4,4,*N*-trimethyl-1,2-thiazetid-3-one 1,1-dioxide gave only the corresponding sulfonic-acid derivative (*via* SO₂ attack) [11]. Heating of **7a** with anhydrous NH₂NH₂ in EtOH led selectively to hydrazide **10** (*via* CO attack) [11]; on the other hand, only sulfonamide **11**⁵⁾ was formed from **9a** in liquid NH₃ (*Scheme 3*). The latter transformation proceeded *via* nucleophilic attack at the SO₂ group and subsequent cleavage of the S–N bond.

The regioselectivity of the nucleophilic ring opening of derivatives of type **7** and **9** has not been investigated systematically so far, and no explanation for the observed selectivities has been given (*cf.* [11]). Therefore, this problem will be the topic of further studies.

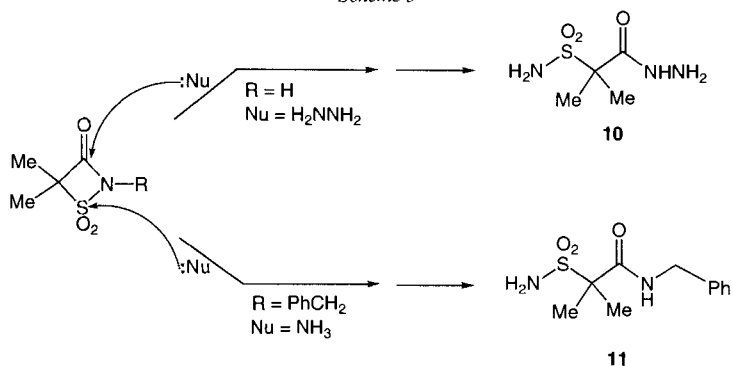
We thank the analytical units of our institute for spectra and analyses, and the *Swiss National Science Foundation* and *F. Hoffmann-La Roche AG*, Basel, for financial support.

³⁾ Benzylation of **7a** to **9a** was performed with benzyl bromide/NaH in DMF (*cf.* also [11] [12]).

⁴⁾ Neither after 24 h at room temperature nor after 1 h in refluxing MeCN was any product detected; refluxing the mixture overnight led to extensive decomposition of the starting materials. Nucleophilic ring-opening reactions of derivatives of type **7** or **9** are known [11].

⁵⁾ The structure of **11** has been established by NMR spectroscopy: the amide NH absorbs at 8.22 ppm as a *t* (*J* = 5.9 Hz) and the SO₂NH₂ group at 6.98 ppm (*s*). The alternative structure with a CONH₂ and a SO₂NHCH₂Ph group was excluded by a long-range H,C-coupling experiment which shows a coupling between PhCH₂ and CO.

Scheme 3



Experimental Part

General. See [2] [3]. If not stated otherwise, IR spectra in KBr, NMR spectra in CDCl_3 at 300 (^1H) and 50.4 (^{13}C) MHz, respectively, and CI-MS with NH_3 .

1. *Starting Materials.* 4,4-Dimethyl-1,2-thiazetid-3-one 1,1-dioxide (**7a**) and 4,4-diethyl-1,2-thiazetid-3-one 1,1-dioxide (**7b**) were prepared according to [11]. Only ^{13}C -NMR and MS data are given; for other data see [11]. **7a**: ^{13}C -NMR ((D_6)DMSO): 165.1 (s, C(3)); 82.2 (s, C(4)); 18.2 (q, $\text{Me}_2\text{C}(4)$). CI-MS: 167 (100, $[\text{M} + \text{NH}_4]^+$), 150 (14, $[\text{M} + 1]^+$). **7b**: ^{13}C -NMR: 164.1 (s, C(3)); 89.7 (s, C(4)); 22.6 (t, 2 MeCH_2); 8.3 (q, 2 MeCH_2). CI-MS: 195 (4, $[\text{M} + \text{NH}_4]^+$), 131 (10), 114 (100). The structure of **7a** was confirmed by X-ray crystallography (Fig. 1 and Table 2). For 3-amino-2H-azirines **1a-d** see refs. in [1].

2. *Reactions of 4,4-Dialkyl-1,2-thiazetid-3-one 1,1-Dioxides 7 with 3-Amino-2H-azirines 1. General Procedure.* To a stirred soln. of **7a** or **7b** in dry MeCN at 0° , a soln. of **1** (1.1 equiv.) in dry MeCN was added and the mixture was left to reach r.t. Workup by CC (SiO_2) or filtration yielded the products as colorless solids which were recrystallized once.

2.1. 3-(Dimethylamino)-4,5,6,7-tetrahydro-4,4,7,7-tetramethyl-1,2,5-thiadiazepin-6-one 1,1-Dioxide (**8a**). 3-(Dimethylamino)-2,2-dimethyl-2H-azirine (**1a**; 62 mg, 0.55 mmol) and **7a** (75 mg, 0.5 mmol) in MeCN (5 ml); $-30^\circ \rightarrow$ r.t. CC with $\text{CH}_2\text{Cl}_2/\text{EtOH}$ 20:3. Recrystallization from DMF/ Et_2O : 40 mg (31%) of **8a**. Colorless crystals. M.p. 256–258°. IR: 3285s, 3200m, 3075m, 3000m, 2980m, 2940m, 1665vs, 1545vs, 1490m, 1465m, 1440m, 1400s, 1380s, 1350m, 1275vs, 1110vs, 1045m, 1005m, 970m, 905m, 855vs, 795m, 730s, 710m, 620s. ^1H -NMR ((D_6)DMSO): 7.58 (s, NH); 3.16 (s, Me_2N); 1.72, 1.42 (2s, $\text{Me}_2\text{C}(4)$, $\text{Me}_2\text{C}(7)$). ^{13}C -NMR ((D_6)DMSO): 170.6, 164.6 (2s, C(3), C(6)); 65.0, 62.1 (2s, C(4), C(7)); 41.9 (q, Me_2N); 29.2, 21.0 (2q, $\text{Me}_2\text{C}(4)$, $\text{Me}_2\text{C}(7)$). CI-MS: 262 (100, $[\text{M} + 1]^+$). Anal. calc. for $\text{C}_{10}\text{H}_{19}\text{N}_3\text{O}_3\text{S}$ (261.35): C 45.96, H 7.33, N 16.08, S 12.27; found: C 46.21, H 7.59, N 16.21, S 12.18.

2.2. 4,5,6,7-Tetrahydro-4,4,7,7-tetramethyl-3-(N-methyl-N-phenylamino)-1,2,5-thiadiazepin-6-one 1,1-Dioxide (**8b**). 2,2-Dimethyl-3-(N-methyl-N-phenylamino)-2H-azirine (**1b**; 383 mg, 2.2 mmol) and **7a** (298 mg, 2 mmol) in MeCN (12 ml); $0^\circ \rightarrow$ r.t., 3 h. Filtration, concentration of the mother liquor and again filtration, recrystallization from $\text{CHCl}_3/\text{EtOH}$: 586 mg (91%) of **8b**. Colorless crystals. M.p. 239.5–242°. IR: 3200w, 3065m, 2980w, 2940w, 1660vs, 1535vs, 1505vs, 1470s, 1380s, 1290vs, 1215w, 1165m, 1150m, 1125vs, 1015w, 1000w, 970m, 930w, 860w, 805s, 780m, 710m, 645w, 620s. ^1H -NMR: 7.5–7.4 (m, 3 arom. H); 7.24 (dd, $J = 6.7, 2.4$, 2 arom. H); 6.57 (s, NH); 3.42 (s, MeN); 1.69, 1.44 (2s, $\text{Me}_2\text{C}(4)$, $\text{Me}_2\text{C}(7)$). ^{13}C -NMR: 175.0, 168.2 (2s, C(3), C(6)); 143.6 (s, 1 arom. C); 129.9, 129.3, 128.0 (3d, 5 arom. CH); 69.8, 66.1 (2s, C(4), C(7)); 47.0 (q, MeN); 31.4, 21.3 (2q, $\text{Me}_2\text{C}(4)$, $\text{Me}_2\text{C}(7)$). CI-MS: 324 (100, $[\text{M} + 1]^+$). Anal. calc. for $\text{C}_{15}\text{H}_{21}\text{N}_3\text{O}_3\text{S}$ (323.42): C 55.71, H 6.55, N 12.99, S 9.92; found: C 55.54, H 6.53, N 13.05, S 10.32.

Crystallization from MeCN/ Et_2O yielded suitable crystals for an X-ray crystal-structure determination (see Fig. 2 and Table 2).

2.3. 8,8-Dimethyl-11-(N-methyl-N-phenylamino)-9-thia-6,10-diazaspiro[4.6]undec-10-en-7-one 9,9-Dioxide (**8c**). 2-(N-Methyl-N-phenylamino)-1-azaspiro[2.4]hept-1-ene (**1c**; 110 mg, 0.55 mmol) and **7a** (75 mg, 0.5 mmol) in MeCN (5 ml); $0^\circ \rightarrow$ r.t. Filtration: 138 mg (79%) of **8c**. Colorless solid. M.p. 280.5–281.5°. IR: 3305w,

3230m, 3090w, 3000w, 2950w, 2870w, 1665vs, 1550vs, 1490s, 1460m, 1390s, 1380s, 1365s, 1295vs, 1205m, 1173s, 1125vs, 1080w, 1050w, 1020w, 1000w, 935m, 910m, 810m, 800s, 790m, 720s, 660w, 620s. ¹H-NMR: 7.5–7.45 (m, 3 arom. H); 7.17 (*d*-like, *J* = 6.7, 2 arom. H); 5.64 (s, NH); 3.41 (s, MeN); 2.2–2.1, 2.0–1.85, 1.7–1.55, 1.5–1.35 (4m, $-(\text{CH}_2)_4-$); 1.78 (s, $\text{Me}_2\text{C}(8)$). ¹³C-NMR: 174.2, 168.8 (2s, C(7), C(11)); 144.8 (s, 1 arom. C); 130.1, 128.7, 127.2 (3*d*, 5 arom. CH); 73.2, 69.0 (2s, C(5), C(8)); 46.5 (*q*, MeN); 42.7, 22.9 (2*t*, $-(\text{CH}_2)_4-$); 22.4 (*q*, $\text{Me}_2\text{C}(8)$). CI-MS: 350 (100, [*M* + 1]⁺). Anal. calc. for C₁₇H₂₃N₃O₃S (349.46): C 58.43, H 6.63, N 12.03, S 9.18; found: C 58.50, H 7.04, N 12.41, S 8.68.

2.4. 4,5,6,7-Tetrahydro-4,7,7-trimethyl-3-(*N*-methyl-*N*-phenylamino)-4-(2-methylpropyl)-1,2,5-thiadiazepin-6-one 1,1-Dioxide (8d). 2-Methyl-3-(*N*-methyl-*N*-phenylamino)-2-(2-methylpropyl)-2H-azirine (1d; 119 mg, 0.55 mmol) and 7a (75 mg, 0.5 mmol) in MeCN (5 ml); 0° → r.t. Evaporation of the solvent, recrystallization from MeCN/Et₂O: 146 mg (80%) of 8d. Colorless crystals. M.p. 228.5–229.5°. IR: 3280s, 2940m, 2860w, 1670vs, 1535vs, 1495vs, 1465m, 1385s, 1350m, 1290vs, 1255m, 1175s, 1180m, 1125vs, 1020w, 1000m, 930m, 900w, 800s, 775m, 740m, 710m, 620s. ¹H-NMR: 7.4–7.35 (m, 3 arom. H); 7.2–7.15 (m, 2 arom. H); 6.31 (s, NH); 3.33 (s, MeN); 2.0–1.8 (m, CH, 1 H of CH₂); 1.64, 1.61 (2s, $\text{Me}_2\text{C}(7)$); 1.30 (*ABM*, *J* = 14.4, 6.1, 1 H of CH₂); 1.25 (s, MeC(4)); 0.90 (*d*, *J* = 6.6, 1 Me of Me₂CH); 0.87 (*d*, *J* = 6.3, 1 Me of Me₂CH). ¹³C-NMR: 173.4, 165.1 (2s, C(3), C(6)); 142.7 (s, 1 arom. C); 128.8, 128.3, 127.3 (3*d*, 5 arom. CH); 68.0, 67.8 (2s, C(4), C(7)); 50.2 (*t*, CH₂); 46.0 (*q*, MeN); 30.4, 22.9, 22.7, 21.0, 19.6 (5*q*, MeC(4), Me₂C(7), Me₂CH); 23.7 (*d*, Me₂CH). CI-MS: 366 (100, [*M* + 1]⁺), 169 (50), 100 (57).

2.5. 3-(Dimethylamino)-7,7-diethyl-4,5,6,7-tetrahydro-4,4-dimethyl-1,2,5-thiadiazepin-6-one 1,1-Dioxide (8e). 1a (84 mg, 0.75 mmol) and 7b (121 mg, 0.68 mmol) in MeCN (5 ml); 0° → r.t. CC with CH₂Cl₂/EtOH 40:3: 118 mg (60%) of 8e. Colorless solid. M.p. 191.5–193°. IR: 3290s, 2965s, 2880w, 1675vs, 1545vs, 1485m, 1445m, 1400s, 1380m, 1355s, 1315w, 1270vs, 1235m, 1155s, 1140s, 1110vs, 1050m, 1020m, 975m, 905m, 875m, 855s, 795m, 765m, 725m, 710m, 690m, 640m, 610m. ¹H-NMR: 6.83 (s, NH); 3.22 (s, Me₂N); 2.2–2.05 (m, 2 MeCH₂); 1.83 (s, Me₂C(4)); 1.09 (*t*, *J* = 7.4, 2 MeCH₂). ¹³C-NMR: 177.7, 171.4 (2s, C(3), C(6)); 81.1, 66.0 (2s, C(4), C(7)); 42.2 (s, Me₂N); 38.6 (*q*, Me₂C(4)); 29.3 (*t*, 2 MeCH₂); 9.1 (*q*, 2 MeCH₂). CI-MS: 290 (100, [*M* + 1]⁺), 226 (16), 192 (16), 155 (36), 98 (21).

2.6. 7,7-Diethyl-4,5,6,7-tetrahydro-4,4-dimethyl-3-(*N*-methyl-*N*-phenylamino)-1,2,5-thiadiazepin-6-one 1,1-Dioxide (8f). 1b (192 mg, 1.1 mmol) and 7b (117 mg, 1 mmol) in MeCN (6 ml); 0° → r.t. Filtration, concentration of the mother liquor, and again filtration, recrystallization from CHCl₃/MeCN: 255 mg (73%) of 8f. Colorless crystals. M.p. 230.5–233.5°. IR: 3300w, 3200m, 3070m, 2990m, 2970m, 2940m, 2880w, 1660vs, 1535vs, 1505s, 1465m, 1400m, 1380s, 1345w, 1295vs, 1215m, 1130vs, 975m, 805s, 780m, 710s, 630s. ¹H-NMR: 7.5–7.45 (m, 3 arom. H); 7.25–7.2 (m, 2 arom. H); 6.78 (s, NH); 3.39 (s, MeN); 2.25–2.1 (m, 2 MeCH₂); 1.39 (s, Me₂C(4)); 1.12 (*t*, *J* = 7.4, 2 MeCH₂). ¹³C-NMR: 177.6, 170.9 (2s, C(3), C(6)); 143.1 (s, 1 arom. C); 129.7, 129.2, 128.0 (3*d*, 5 arom. CH); 81.7, 67.2 (2s, C(4), C(7)); 46.6 (*q*, MeN); 30.2 (*q*, Me₂C(4)); 23.8 (*t*, 2 MeCH₂); 9.1 (*q*, 2 MeCH₂). CI-MS: 352 (100, [*M* + 1]⁺), 288 (18), 218 (11), 183 (12), 108 (64). Anal. calc. for C₁₇H₂₅N₃O₃S (351.48): C 58.09, H 7.17, N 11.96, S 9.12; found: C 57.44, H 7.19, N 11.85, S 8.83.

2.7. 7,7-Diethyl-4,5,6,7-tetrahydro-4-methyl-3-(*N*-methyl-*N*-phenylamino)-4-(2-methylpropyl)-1,2,5-thiadiazepin-6-one 1,1-Dioxide (8g). 1d (119 mg, 0.55 mmol) and 7b (89 mg, 0.5 mmol) in MeCN (5 ml); 0° → r.t., 1 h. CC with hexane/acetone/AcOEt 4:1:1: 159 mg (81%) of 8g. Colorless solid. M.p. 173.5–174.5°. IR: 3300w, 3200m, 3080m, 2980s, 2960s, 2875m, 1660vs, 1545vs, 1495s, 1470s, 1430m, 1390s, 1340m, 1295vs, 1255m, 1190m, 1165s, 1120vs, 1075m, 1020m, 1000m, 950w, 930w, 895w, 800s, 775m, 730m, 710m, 700m, 615s. ¹H-NMR: 7.5–7.45 (m, 3 arom. H); 7.25–7.2 (m, 2 arom. H); 6.63 (s, NH); 3.38 (s, MeN); 2.3–2.1 (m, 2 MeCH₂); 2.1–2.0 (m, 1 H of CH₂); 1.9–1.8 (m, CH); 1.31 (s, MeC(4)); 1.35–1.25 (m, 1 H of CH₂); 1.14, 1.11 (2*t*, *J* = 7.5, 2 MeCH₂); 0.95, 0.83 (2*d*, *J* = 6.7, Me₂CH). ¹³C-NMR: 176.8, 170.3 (2s, C(3), C(6)); 143.5 (s, 1 arom. C); 129.7, 129.3, 128.4 (3*d*, 5 arom. CH); 80.8, 70.6 (2s, C(4), C(7)); 49.6 (*t*, CH₂); 46.9 (*q*, MeN); 29.1 (*q*, MeC(4)); 24.4, 23.9 (2*q*, Me₂CH); 24.3, 24.1 (2*t*, 2 MeCH₂); 9.3, 9.1 (2*q*, 2 MeCH₂). CI-MS: 394 (100, [*M* + 1]⁺), 197 (34), 100 (83). Anal. calc. for C₂₀H₃₁N₃O₃S (393.56): C 61.04, H 7.94, N 10.68, S 8.15; found: C 61.35, H 7.96, N 10.91, S 8.74.

3. Reactions of 2-Benzyl-4,4-dimethyl-1,2-thiazetidin-3-one 1,1-Dioxide (9a). 3.1. With 1a. A soln. of 1a (15 mg, 0.13 mmol) and 9a (24 mg, 0.1 mmol) in dry MeCN (3 ml) was stirred at r.t. No reaction was observed after 24 h, nor after 1 h at 81° (reflux). Refluxing the mixture overnight (ca. 15 h) led to an intractable mixture of decomposed material.

3.2. With NH₃. To 36 mg (0.15 mmol) of 9a at –78°, liquid NH₃ (10 ml) was slowly added. The soln. was stirred without cooling until reaching r.t. The remaining colorless oil was treated with Et₂O and the solid material recrystallized from Et₂O: 35 mg (91%) *N*-Benzyl-2-methyl-2-sulfamoylpropanamide (11). Colorless crystals. M.p. 103–106°. ¹H-NMR ((D₆)DMSO): 8.22 (*t*, *J* = 5.90, NH); 7.35–7.15 (m, 5 arom. H); 6.98 (s, NH₂); 4.33 (*d*-like, PhCH₂); 1.52 (s, Me₂C). ¹³C-NMR ((D₆)DMSO): 168.3 (s, CO); 139.2 (s, 1 arom. C); 128.1, 126.7, 126.5 (3*d*,

5 arom. CH); 65.7 (s, Me₂C); 42.7 (t, PhCH₂); 21.2 (s, Me₂C). A long-range H,C-coupling between PhCH₂ and CO was established (CDCl₃, 298 K, coloc puls program). CI-MS: 274 (100, [M + NH₄]⁺), 257 (30, [M + 1]⁺), 176 (12).

4. *Crystal-Structure Determination of 7a and 8b* (see Table 2 and Figs. 1 and 2)⁶⁾. The intensities were collected on a Rigaku-AFC5R diffractometer using graphite-monochromated MoK_α radiation ($\lambda = 0.71069 \text{ \AA}$) and a 12 kW rotating anode generator. The intensities were corrected for Lorentz and polarization effects, but not for absorption. Data collection and refinement parameters are listed in Table 2, views of the molecules and packing diagrams are shown in Figs. 1 and 2. The structures were solved by direct methods using SHELXS86 [14], which revealed the positions of all non-H-atoms. The non-H-atoms were refined anisotropically. All of the H-atoms were located in difference electron density maps, and their positions were allowed to refine together with individual isotropic temperature factors. Refinements of the structures were carried out on *F* using full-matrix least-squares procedures. Neutral-atom scattering factors for non-H-atoms were taken from [15a] and the scattering factors for H-atoms from [16]. Anomalous dispersion effects were included in *F_c* [17]; the values for *f'* and *f''* were those of [15b]. All calculations were performed using the TEXSAN [18] crystallographic software package.

Table 2. Crystallographic Data for Compounds 7a and 8b

	7a	8b
Crystallized from	EtOH/pentane	MeCN/Et ₂ O
Empirical formula	C ₄ H ₇ NO ₃ S	C ₁₅ H ₂₁ N ₃ O ₃ S
Formula weight	149.16	323.41
Crystal color, habit	colorless, irreg. prism	colorless, prism
Crystal dimensions [mm]	0.38 × 0.38 × 0.38	0.20 × 0.23 × 0.50
Crystal temp. [K]	173(1)	173(1)
Crystal system	monoclinic	monoclinic
Space group	<i>C2/c</i>	<i>P2₁/n</i>
Z	8	4
Lattice parameters		
Reflections for cell determination	25	25
2θ range [°]	39 < 2θ < 40	38 < 2θ < 40
<i>a</i> [Å]	18.689(6)	12.077(3)
<i>b</i> [Å]	8.347(5)	10.962(4)
<i>c</i> [Å]	9.061(4)	12.852(4)
β [°]	113.88(2)	93.67(2)
<i>V</i> [Å ³]	1292.6(9)	1698.1(8)
<i>D_x</i> [g cm ⁻³]	1.533	1.265
Absorption coefficient μ (MoK _α) [mm ⁻¹]	0.433	0.206
Scan type	ω/2θ	ω/2θ
2θ(max) [°]	60	60
Total reflections measured	2066	5408
Symmetry-independent reflections	1896	4945
Reflections observed [<i>I</i> > 2σ(<i>I</i>)]	1663	3510
Variables	110	283
Final <i>R</i>	0.0315	0.0439
<i>R_w</i> ^{a)}	0.0358	0.0393
Goodness of fit <i>s</i>	2.849	1.756
Final <i>A</i> _{max} /σ	0.0002	0.0006
Δρ (max, min) [e Å ⁻³]	0.31, -0.50	0.47, -0.39

^{a)} Function minimized $\sum w(|F_o| - |F_c|)^2$, $1/w = \sigma^2(F_o) + (0.005F_o)^2$.

⁶⁾ Crystallographic data (excluding structure factors) for the structures reported in this paper have been deposited with the Cambridge Crystallographic Data Centre as supplementary publication No. CCDC-10/28. Copies of the data can be obtained, free of charge, on application to the director, CCDC, 12 Union Road, Cambridge CB2 1EZ, UK (fax: +44 (0)1223 336033 or e-mail: teched@chemcrs.cam.ac.uk).

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